

Supplemental Information

Reworking the *Tao–Mo* exchange-correlation functional:

I. Reconsideration and Simplification

H. Francisco

Quantum Theory Project, Dept. of Physics, University of Florida, Gainesville FL 32611

A.C. Cancio

Center for Computational Nanoscience, Dept. of Physics and Astronomy, Ball State University, Muncie IN 47306

S.B. Trickey

*Quantum Theory Project, Dept. of Physics and Dept. of Chemistry, University of Florida, Gainesville FL 32611**

(Dated: 15 July 2023; revised 21 Sept. 2023)

We provide detailed equations for the *rreg* TM C functional, information on the *v1-sreg* TM version of our simplified X functional; a careful derivation of the gradient expansion of *sreg*TM exchange energy and how to satisfy the correct gradient expansion; lastly a system-by-system tabulation of the numerical results, including *v1-sreg*TM and *v3-sreg*TM, testing against standard molecular and crystalline system data sets.

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* francisco.hector@ufl.edu; trickey@ufl.edu

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I. CORRELATION FUNCTIONAL DETAILS

The *rreg* C energy per particle is given by

$$\epsilon_c^{rregTM} := \epsilon_c^0 f_{c1}(\alpha) + \epsilon_c^1 f_{c2}(\alpha) , \quad (1)$$

The interpolation functions f_{c1} , f_{c2} are defined below.

Quantities used in the various contributions are defined as

$$\zeta := (n_\uparrow - n_\downarrow)/(n_\uparrow + n_\downarrow) \quad (2)$$

and

$$\xi := \nabla \zeta / 2(3\pi^2)^{2/3} . \quad (3)$$

Here σ is the spin label, $\sigma = \uparrow$ or $\sigma = \downarrow$, and $n_\sigma = \sum_i |\varphi_{i\sigma}|^2$ (assuming integer occupation for simplicity).

The correlation energy per particle ingredients are

$$\epsilon_c^0 := (\epsilon_c^{LDA0} + H_0^{SCAN})G_c(\zeta) \quad (4)$$

and

$$\epsilon_c^1 := \epsilon_c^{LSDA1} + H_1^{PBE} . \quad (5)$$

Here ϵ_c^{LSDA1} is the PW92 local spin density approximation (LSDA) correlation energy density [1] and H_1^{PBE} is

$$H_1^{PBE} = \gamma \phi^3 \ln[1 + w_1(1 - g_1(At^2))] . \quad (6)$$

Definitions of the quantities involved are

$$t = (3\pi^2/16)^{1/3} s / \phi r_s^{1/2} \quad (7)$$

$$A = \beta(r_s) / (\gamma w_1) \quad (8)$$

$$w_1 = \exp[-\epsilon_c^{LSDA1} / (\gamma \phi^3)] - 1 \quad (9)$$

$$g_1(At^2) = 1 / (1 + At^2 + A^2 t^4) \quad (10)$$

$$\beta(r_s) = 0.066725(1 + 0.1r_s) / (1 + 0.1778r_s) \quad (11)$$

$$\phi(\zeta) = [(1 + \zeta)^{2/3} + (1 - \zeta)^{2/3}] / 2 . \quad (12)$$

As usual, the local Wigner-Seitz radius is a function of the density $n(\mathbf{r})$,

$$r_s(\mathbf{r}) = (4\pi n(\mathbf{r})/3)^{-1/3} . \quad (13)$$

Also $\gamma = 0.031091$.

In Eq. 4, $\epsilon_c^{LDA0} G_c(\zeta)$ is the local correlation energy density form from the SCAN functional [2, 3]

$$\epsilon_c^{LDA0} = \frac{-b_{1c}}{1 + b_{2c}r_s^{1/2} + b_{3c}r_s} , \quad (14)$$

and

$$\begin{aligned} G_c(\zeta) &= \{1 - 2.3631(d_x(\zeta) - 1)\}(1 - \zeta^{12}) \\ d_x(\zeta) &= [(1 + \zeta)^{4/3} + (1 - \zeta)^{4/3}]/2 , \end{aligned} \quad (15)$$

and

$$H_0^{SCAN} = b_{1c} \ln[1 + w_0(1 - g_\infty(\zeta = 0, s))] . \quad (16)$$

The various quantities are

$$w_0 := \exp[-\epsilon_c^{LDA0}/b_{1c}] - 1 \quad (17)$$

$$g_\infty(\zeta, s) := \lim_{r_s \rightarrow \infty} g_0(At^2) = 1/(1 + 4\chi_\infty s^2)^{1/4} \quad (18)$$

$$\chi_\infty(\zeta) := (3\pi^2/16)^{2/3} \beta(r_s \rightarrow \infty) \phi/[C_x(\zeta) - f_0] \quad (19)$$

$$C_x(\zeta) = (-3/4\pi)(9\pi/4)^{1/3} d_x(\zeta) . \quad (20)$$

$f_0 = -0.9$, at $\zeta = 0$, $\chi_\infty(\zeta = 0) = 0.128026$. The parameters are $b_{1c} = 0.0285764$, $b_{2c} = 0.0889$, and $b_{3c} = 0.125541$.

Finally, the interpolation functions $f_{c1}(\alpha)$ and $f_{c2}(\alpha)$ that appear in Eq. (1) are

$$f_{c2}(\alpha) = \frac{3[g(\alpha)]^3}{1 + [g(\alpha)]^3 + [g(\alpha)]^6} , \quad (21)$$

with

$$g(\alpha) = \frac{(1 + \gamma_1)\alpha}{\gamma_1 + \alpha} , \quad \gamma_1 = 0.2, \quad (22)$$

and

$$f_{c1}(\alpha) = 1 - f_{c2}(\alpha) . \quad (23)$$

The quantity $\tilde{\epsilon}_c = \max[\epsilon_c^{PBE}(n_\sigma, 0, \nabla n_\sigma, 0), \epsilon_c^{PBE}(n_\uparrow, n_\downarrow, \nabla n_\uparrow, \nabla n_\downarrow)]$, where ϵ_c^{PBE} is the PBE GGA correlation energy per particle [4].

In original TM, instead of the TPSS C function

$$C(\zeta, \xi)^{TPSS} = \frac{0.53 + 0.87\zeta^2 + 0.50\zeta^4 + 2.26\zeta^6}{\{1 + \xi^2[(1 + \zeta)^{-4/3} + (1 - \zeta)^{-4/3}]/2\}^4} , \quad (24)$$

C is given by

$$C(\zeta, \xi)^{TM} = \frac{0.10\zeta^2 + 0.32\zeta^4}{\{1 + \xi^2[(1 + \zeta)^{-4/3} + (1 - \zeta)^{-4/3}]/2\}^4} . \quad (25)$$

II. ANALYSIS OF NEARLY UNIFORM ELECTRON GAS LIMIT OF *sreg*TM META-GGA FUNCTIONALS.

A. General structure of TM in the nearly-uniform limit.

The slowly varying electron gas is defined by $p, |q| \ll 1$, with p as defined in Eq. (5) of the text and $q := \nabla^2 n / 4k_F^2 n$ ($k_F := (3\pi^2 n)^{1/3}$) is the reduced Laplacian of the density. In this regime α has the gradient expansion

$$\alpha \rightarrow 1 + \frac{20}{9}q - \frac{40}{27}p. \quad (26)$$

The homogeneous electron gas limit is the extreme case $p = q = 0$ and $\alpha = 1$.

From the definitions of w Eq. (19) and z_{rev} , Eq. (27) of the main text, their homogeneous electron gas limits in terms of the regularization parameter ϵ_p are

$$z_0(\epsilon_p) := z_{rev}(p = 0, \alpha = 1) = \frac{\epsilon_p}{3 + \epsilon_p} \quad (27)$$

$$w_0(\epsilon_p) := w(z_0) = \frac{z_0^2 + 3z_0^3}{(1 + z_0^3)^2}. \quad (28)$$

For $\epsilon_p = 0.5$ these go to

$$z_{rev} \rightarrow z_0(0.5) = 1/7 \quad (29)$$

$$w \rightarrow w_0(0.5) = 0.02899. \quad (30)$$

The essential structure of the TM enhancement factor F_x is

$$F_x = wF_x^{DME} + (1 - w)F_x^{SC}. \quad (31)$$

Since the two limiting functionals for exchange (Eq. (7) and (17) of the main text) both tend to one in the homogeneous electron gas limit, namely

$$F_x^{SC} \rightarrow 1 \quad (32)$$

$$F_x^{DME} \rightarrow 1, \quad (33)$$

the limit of Eq. (31) becomes

$$F_x^{(0)} = w_0 + (1 - w_0) = 1 \quad (34)$$

irrespective of the value of w_0 , and thus of ϵ_p . Thus the homogeneous electron gas limit is correct for all variants of *sreg*TM exchange.

For the weakly varying case, $p, q \rightarrow 0$, the X enhancement factor may be expressed as

$$F_x \sim 1 + F_x^{(2)} + F_x^{(4)} + \dots. \quad (35)$$

Here (0) indicates the value for uniform density, (2) the second-order gradient expansion correction, i.e., the term linear in p and q , and (4) is the correction for the fourth-order term in the gradient expansion, with terms proportional to p^2 , q^2 , and pq .

To get the second-order result, we need expansions for the quantities in Eq. (31). We start by simply proposing

gradient expansions for w , F_x^{DME} and F_x^{SC} :

$$w \sim w_0 + w_p p + w_q q \quad (36)$$

$$F_x^{SC} \sim 1 + \delta_p^{SC} p + \delta_q^{SC} q \quad (37)$$

$$F_x^{DME} \sim 1 + \delta_p^{DME} p + \delta_q^{DME} q \quad (38)$$

It will not be necessary to evaluate $w_{p,q}$, but the others must be done. Substituting those three expressions into Eq. (31) and taking all terms to second order gives

$$\begin{aligned} F_x^{(2)} &= (w_p p + w_q q) + w_0 (\delta_p^{DME} p + \delta_q^{DME} q) \\ &\quad + (1 - w_0) (\delta_p^{SC} p + \delta_q^{SC} q) - (w_p p + w_q q) . \end{aligned} \quad (39)$$

That has the simple form

$$F_x^{(2)} = \delta_p^{SC} p + \delta_q^{SC} q + w_0 (\Delta_p p + \Delta_q q) , \quad (40)$$

in which we have defined

$$\Delta_p := \delta_p^{DME} - \delta_p^{SC} \quad (41)$$

$$\Delta_q := \delta_q^{DME} - \delta_q^{SC} . \quad (42)$$

It is important to observe that the structure of the various TM X functionals has the DME (“density matrix expansion”) and SC (“slowly varying correction” per Ref. 5) contributions. Therefore we expect the desired quantity to be the gradient expansion for the SC term, since it is designed to return the correct one. That is, the error in our functional should be characterized by Δ_p and Δ_q . Unfortunately, the SC itself has errors in the gradient expansion in sregTM, caused by a nonzero value of z for $p, q = 0$. Since the desired gradient expansion correction is

$$F_x^{(2)} = \frac{10}{81} p \quad (43)$$

our error expression becomes

$$\delta F_x^{(2)} = (\delta_p^{SC} - 10/81) p + \delta_q^{SC} q + w_0 (\Delta_p p + \Delta_q q) . \quad (44)$$

Note that the magnitude of the error is of order w_0 , which is a small number, and we expect an error of a few percent.

B. Working out gradient expansion error

1. The “SC” term

The form for F_x^{sc} is given in Eq. (17) of the main text. It depends upon $\tilde{q}$, which is designed to be a substitute for q , hence

$$\tilde{q} \rightarrow q \quad (45)$$

in the slowly-varying limit. Given this, the second-order term in the gradient expansion of Eq. (17) is

$$\frac{10}{81} p - \frac{73}{405} q \frac{3}{5} z_0 (1 - z_0) \quad (46)$$

which, after substitutions is

$$\delta^{SC} = \frac{10}{81}p - \frac{73}{225} \frac{\epsilon_p}{(3 + \epsilon_p)^2} q, \quad (47)$$

or

$$\delta_p^{SC} = \frac{10}{81} \quad (48)$$

$$\delta_q^{SC} = -\frac{73}{225} \frac{\epsilon_p}{(3 + \epsilon_p)^2}. \quad (49)$$

The latter coefficient reduces to -146/11025 or roughly 0.013 if $\epsilon_p = 0.5$, a factor of ten smaller than the exact coefficient 10/81. The second-order error Eq. (44) now is reduced to

$$\delta F_x^{(2)} = \delta_q^{SC} + w_0(\Delta_p p + \Delta_q q). \quad (50)$$

2. The “DME” term

Analysis of the contribution from this term is much harder. Begin with the definitions for F_x^{DME} , Eq. (7), f , Eq. (8), and R , Eq. (9) in the main text, along with the associated constants $\beta = 79.873$, $\lambda = 0.6866$ and auxiliary functions

$$y := (2\lambda - 1)^2 p \quad (51)$$

$$z_2 := 5p/3 \quad (52)$$

$$z_3 := z_2 + \alpha \quad (53)$$

$$z_4 := z_3 - 1 - z_2/9. \quad (54)$$

We have defined z_4 to make later analysis easier. The other three were defined at Eqs. (10), (11), and (12) of the main text.

Working backwards, the slowly varying limits of these are

$$y \rightarrow (2\lambda - 1)^2 p \quad (55)$$

$$z_2 \rightarrow \frac{5}{3}p \quad (56)$$

$$z_3 \rightarrow 1 + \frac{20}{9}q + \frac{5}{27}p \quad (57)$$

$$z_4 \rightarrow \frac{20}{9}q \quad (58)$$

and

$$\begin{aligned} R \sim & 1 + 595(2\lambda - 1)^2 \frac{p}{54} - \left(1 + \frac{20}{9}q + \frac{5}{27}p\right) \\ & + [3(\lambda^2 - \lambda + 1/2)] \frac{20}{9}q \end{aligned} \quad (59)$$

or

$$\left[\frac{595}{54}(2\lambda - 1)^2 - \frac{5}{27} \right] p + [3(\lambda^2 - \lambda + 1/2) - 1] \frac{20}{9}q \quad (60)$$

Also

$$f \rightarrow 1 + \frac{70}{27}y \quad (61)$$

or

$$f \rightarrow 1 + \frac{70}{27}(2\lambda - 1)^2 p \quad (62)$$

so that

$$f^{-n} \rightarrow 1 - n \frac{70}{27}(2\lambda - 1)^2 p \quad (63)$$

Combined, these yield

$$F_x^{DME} \rightarrow 1 - 2 \frac{70}{27}(2\lambda - 1)^2 p + \frac{7}{9} R \quad (64)$$

so that we get second-order gradient expansion coefficients that read

$$\begin{aligned} \delta_p^{DME} &= -\frac{140}{27}(2\lambda - 1)^2 \\ &\quad + \frac{7}{9} \left[\frac{595}{54}(2\lambda - 1)^2 - \frac{5}{27} \right] \end{aligned} \quad (65)$$

$$\delta_q^{DME} = [3(\lambda^2 - \lambda + 1/2) - 1] \frac{140}{81} . \quad (66)$$

With $\lambda = 0.6866$, these evaluate to $\delta_p^{DME} = 0.32739$ and $\delta_q^{DME} = -0.25155$.

3. Putting it all together

Overall, for $\delta F_x^{(2)}$ from Eqs. 50, 41, 42, 49, 65 and 66 we get

$$\delta F_x^{(2)} = \delta_q^{SC} + w_0 \left[\left(\delta_p^{DME} - \frac{10}{81} \right) p + (\delta_q^{DME} - \delta_q^{SC}) q \right] . \quad (67)$$

To evaluate these coefficients we used a python script to get

$$\delta F_x^{(2)} = \delta_p p + \delta_q q . \quad (68)$$

For the chosen value $\epsilon_p = 0.5$, the result is $\delta_p = 0.005912$ and $\delta_q = -0.020151$.

If one wishes, using a gauge invariant transformation of the term dependent on q to one depending on p ,

$$\delta_q q \rightarrow \delta_q p/3 \quad (69)$$

using, via integration by parts,

$$\int q e_x^{LDA} d^3 r = \frac{1}{3} \int p e_x^{LDA} d^3 r \quad (70)$$

where $e_x^{LDA} \sim n^{4/3}$ is LDA (Dirac) exchange energy density. This yields a final estimate of the gradient expansion error in $v2$ -sregTM of

$$\delta F_{x,v2}^{(2)} = (\delta_p + \delta_q/3)p = -0.000805p. \quad (71)$$

Fortunate error cancellation puts the result within 0.6% error of the correct value of 0.1235, in practice therefore not leading to the need for any meaningful correction.

The fortuitous cancellation of errors in $v2$ -sregTM between error arises from from the z -dependence of F_x^{SC} and of

w . Noting that both terms are of course functions of ϵ_p , we have an overall function of ϵ_p , $\delta F_x^{(2)}(\epsilon_p)$ that is near a root for $\epsilon = 0.5$. A numerical search for this root returns:

$$\epsilon_p^{GE2} = 0.58568 \quad (72)$$

which then yields a fractional error in the second-order contribution to the exchange energy of -2×10^{-10} .

Finally we note that for $v1-sreg$ TM, there are changes to z_{rev} only in w and none in F_x^{SC} . Taking this into account sets δ_q^{SC} to zero everywhere and changes Eq. 67 to

$$\delta F_{x,v1}^{(2)} = w_0 \left[\left(\delta_p^{DME} - \frac{10}{81} \right) p + \delta_q^{DME} q \right] \quad (73)$$

with the net result after integration by parts of

$$\delta F_{x,v1}^{(2)} = 0.003481p. \quad (74)$$

This is still only a few percent off the correct answer, but we lack an obvious mechanism for enforcing exact compliance.

III. $v1-sreg$ TM X

Here we provide parametrization for the $v1-sreg$ TM X functional and comparison of its performance with the $v2-sreg$ TM X functional when used with both $rreg$ TM and PBE correlation functionals.

TABLE I. Molecular test set results for the $v1-sreg$ TM exchange functional combined with the $rreg$ TM correlation functional for different values of ϵ_p . Heat of formation errors in $kcal/mol$, bond length errors in $\text{\AA}$, and frequency errors in cm^{-1} .

	ϵ_p	$v1-sreg$ TM X + $rreg$ TM C							$rreg$ TM XC
		0.01	0.1	0.2	0.3	0.4	0.5	0.6	
Heats of Formation	ME	3.228	1.953	0.501	-0.905	-2.233	-3.467	-4.604	-3.790
	MAE	8.706	7.729	6.860	6.258	5.923	5.885	6.194	5.612
Bonds	ME	0.015	0.015	0.015	0.014	0.014	0.013	0.011	0.012
	MAE	0.018	0.017	0.017	0.016	0.016	0.015	0.014	0.014
Frequencies	ME	-25.941	-24.788	-23.531	-22.190	-20.841	-19.618	-18.388	-21.011
	MAE	36.649	36.011	35.490	34.998	34.608	34.28	33.957	35.578

Table II gives the results for both $v1-sreg$ TM X and $v2-sreg$ TM X for the molecular test sets.

Table III gives the results for both $v1-sreg$ TM X and $v2-sreg$ TM X for the solid test sets.

Note that the use of $v1-sreg$ TM X instead of $v2-sreg$ TM X degrades bulk modulus performance for solids, but otherwise is comparable with the second variant $v2-sreg$ TM X. Since, however, the second variant is simpler and slightly better in overall performance, we chose it.

IV. MAGNETIZATION ENERGETICS

Table IV gives the saturation magnetic moment results for both $v1-sreg$ TM X and $v2-sreg$ TM X for the $3d$ solids in comparison with other functionals.

Details follow in Tables V, VI, and VII.

TABLE II. Summary of molecular test set results for the $v1$ -sregTM and $v2$ -sregTM xchange functionals (both with $\epsilon_p = 0.5$) combined with the $rreg$ TM correlation functional, as well as $v2$ -sregTM X with PBE C versus $rreg$ TM X combined with PBE C, all compared with PBE XC and r²SCAN XC. Heat of formation errors in $kcal/mol$, bond length errors in $\text{\AA}$, and frequency errors in cm^{-1} .

	X C	$rreg$ TM $rreg$ TM	$v1$ -sregTM $rreg$ TM	$v2$ -sregTM $rreg$ TM	$v3$ -sregTM $rreg$ TM	$rreg$ TM PBE	$v2$ -sregTM PBE	r ² SCAN r ² SCAN	PBE PBE
Heats of Formation	ME	-3.79	-3.467	-3.512	-4.411	-2.208	-1.520	-3.145	-20.878
	MAE	5.612	5.885	5.895	6.128	5.452	5.354	4.488	21.385
Bonds	ME	0.012	0.013	0.013	0.011	0.008	0.008	0.005	0.018
	MAE	0.014	0.015	0.015	0.014	0.011	0.011	0.010	0.018
Frequencies	ME	-21.011	-19.618	-19.275	-18.354	-4.667	-2.865	11.336	-33.781
	MAE	35.578	34.28	34.272	34.120	25.183	24.138	30.899	43.613

TABLE III. Comparison of solid system errors for DFA combinations as in Table II for four solid test sets. Equilibrium lattice constant errors in $\text{\AA}$, cohesive energy errors in $eV/atom$, bulk modulus errors in GPa , and Kohn-Sham band gap errors in eV .

	X C	$rreg$ TM $rreg$ TM	$v1$ -sregTM $rreg$ TM	$v2$ -sregTM $rreg$ TM	$v3$ -sregTM $rreg$ TM	$rreg$ TM PBE	$v2$ -sregTM PBE	r ² SCAN r ² SCAN	PBE PBE
Lattice constants	ME	0.000	0.002	0.004	0.005	-0.006	-0.002	0.026	0.046
	MAD	0.029	0.030	0.031	0.031	0.028	0.029	0.037	0.053
Cohesive energies	ME	0.212	0.180	0.159	0.148	0.248	0.199	-0.134	-0.070
	MAD	0.251	0.232	0.216	0.210	0.288	0.251	0.238	0.252
Bulk moduli	ME	1.856	0.666	0.223	0.140	4.375	2.732	1.367	-9.704
	MAD	6.740	6.811	6.602	6.450	7.200	6.542	5.963	11.022
Band Gaps	ME	-1.52	-1.54	-1.53	-1.52	-1.42	-1.44	-1.20	-1.69
	MAD	1.52	1.54	1.53	1.52	1.42	1.44	1.20	1.69

TABLE IV. Saturation magnetic moments in μ_B for some $3d$ solids with $v1$ -sregTM and $v2$ -sregTM X functionals (both with $\epsilon_p = 0.5$) combined with the $rreg$ TM correlation functional, as well as $v2$ -sregTM X with PBE C versus $rreg$ TM X combined with PBE C, all compared with PBE XC and r^2 SCAN XC. “Exp.” denotes the experimental saturation magnetization.

	Exp.	X C	$rreg$ TM $rreg$ TM	$v1$ -sregTM $rreg$ TM	$v2$ -sregTM $rreg$ TM	$rreg$ TM PBE	$v2$ -sregTM PBE	r^2 SCAN r^2 SCAN	PBE PBE
Fe	2.22		2.10	2.16	2.17	2.10	2.18	2.64	2.18
Co	1.72		1.72	1.73	1.73	1.72	1.74	1.83	1.64
Ni	0.62		0.68	0.66	0.66	0.68	0.66	0.72	0.63

TABLE V. Energy of bcc Fe and its saturation magnetic moments for the different functionals sorted in descending order. “Exp.” denotes the experimental saturation magnetization.

Functional	Energy eV	Magnetization μ_B
Exp.		2.22
PBE XC	-0.5293	2.18
$rreg$ TM X + PBE C	-0.6516	2.10
$rreg$ TM XC	-0.6535	2.10
$v1$ -sregTM X + $rreg$ TM C	-0.6653	2.16
$v2$ -sregTM X + $rreg$ TM C	-0.6751	2.17
$v2$ -sregTM X + PBE C	-0.6770	2.18
r^2 SCAN XC	-1.2375	2.64

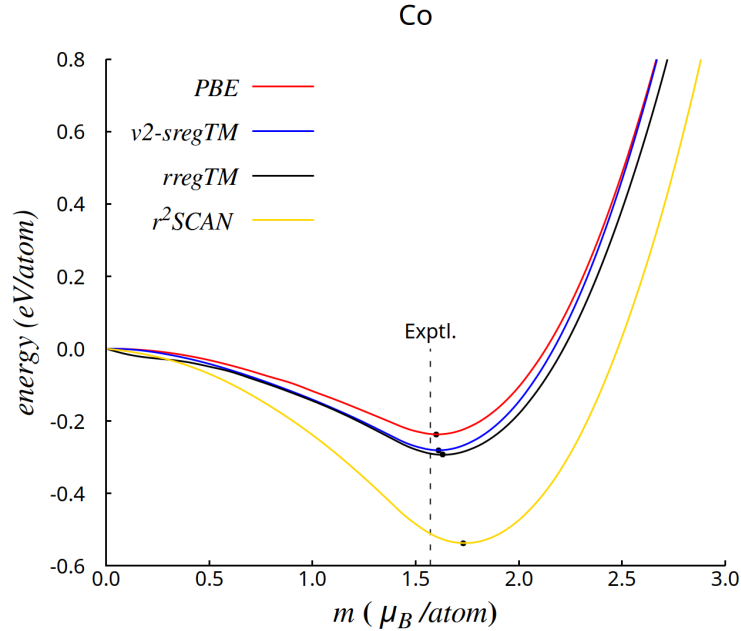


FIG. 1. Fixed spin moment energy for hcp Co using calculated equilibrium lattice parameter, on a per-atom basis for. Comparison of different functionals.

TABLE VI. As in Table V for fcc Co.

Functional	Energy eV	Magnetization μ_B
Exp.		1.72
PBE XC	-0.2055	1.64
<i>v1-sreg</i> TM X + <i>rreg</i> TM C	-0.2875	1.73
<i>v2-sreg</i> TM X + <i>rreg</i> TM C	-0.2946	1.73
<i>rreg</i> TM XC	-0.2996	1.72
<i>v2-sreg</i> TM X + PBE C	-0.3032	1.74
<i>reg</i> TM X + PBE C	-0.3082	1.72
r ² SCAN XC	-0.5599	1.83

TABLE VII. As in Table V for fcc Ni.

Functional	Energy eV	Magnetization μ_B
Exp.		0.62
PBE XC	-0.0583	0.63
<i>v1-sreg</i> TM X + <i>rreg</i> TM C	-0.0805	0.66
<i>v2-sreg</i> TM X + PBE C	-0.0811	0.66
<i>v2-sreg</i> TM X + <i>rreg</i> TM C	-0.0814	0.66
<i>rreg</i> TM XC	-0.0875	0.68
<i>reg</i> TM X + PBE C	-0.0879	0.68
r ² SCAN XC	-0.1187	0.72

TABLE VIII. As in Table V for hcp Co.

Functional	Energy eV	Magnetization μ_B
Exp.		1.57
PBE XC	-0.2369	1.60
<i>v2-sreg</i> TM X + <i>rreg</i> TM C	-0.2808	1.61
<i>rreg</i> TM XC	-0.2934	1.63
<i>reg</i> TM X + PBE C	-0.3002	1.63
<i>v1-sreg</i> TM X + <i>rreg</i> TM C	-0.3052	1.63
<i>v2-sreg</i> TM X + PBE C	-0.3164	1.64
r ² SCAN XC	-0.5375	1.73

V. MOLECULAR TEST SET DETAILED RESULTS

Here we provide the molecule-by-molecule tabulation for the G3/99, T96R, and T82-F test sets for *rreg*TM compared with *sreg*TM and, r²SCAN, and PBE.

TABLE IX: Standard enthalpies of formation (ΔH_f) in kcal/mol for the G3X/99 molecular test set [6, 7] obtained with different approximate exchange-correlation functionals.

Molecule	Exp.	X	<i>rreg</i> TM	<i>v1-sreg</i> TM	<i>v2-sreg</i> TM	<i>v3-sreg</i> TM	<i>rreg</i> TM	<i>v2-sreg</i> TM	r ² SCAN	<i>PBE</i>
		C	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	<i>PBE</i>	<i>PBE</i>	r ² SCAN	<i>PBE</i>
LiH	33.30		35.971	35.928	35.926	35.945	38.861	38.815	36.369	38.353
² BeH	81.70		76.928	77.007	77.000	77.005	74.458	74.531	72.401	77.032
² CH	142.50		141.528	141.970	141.954	141.938	140.283	140.695	145.246	141.870
CH ₂ (³ B ₁)	93.60		91.154	90.563	90.538	90.318	92.444	91.891	87.380	89.512
CH ₂ (¹ A ₁)	102.60		106.085	107.298	107.274	107.033	104.267	105.401	109.080	104.969
³ CH ₃	35.00		33.582	34.330	34.291	33.897	34.199	34.936	31.093	32.400
CH ₄	-17.80		-16.031	-14.334	-14.357	-14.887	-15.474	-13.697	-16.985	-18.051
³ NH	85.80		77.885	78.929	78.907	78.911	78.469	79.532	84.819	80.598
NH ₂ (² B ₁)	44.50		36.237	38.863	38.827	38.547	37.917	40.560	43.940	38.882
NH ₃	-10.90		-13.125	-10.482	-10.516	-11.007	-9.982	-7.192	-6.591	-14.099
² OH	9.00		7.318	8.397	8.412	8.140	9.887	11.021	7.906	6.631
H ₂ O	-57.80		-51.789	-51.939	-51.885	-52.383	-46.419	-46.389	-54.083	-58.172
HF	-65.20		-57.560	-59.793	-59.707	-59.918	-53.818	-55.930	-61.578	-65.067
SiH ₂ (¹ A ₁)	65.20		67.435	67.561	67.616	67.642	64.694	64.883	67.659	68.485
SiH ₂ (³ B ₁)	86.20		82.278	82.201	82.261	82.192	80.983	81.012	78.683	84.892
² SiH ₃	47.90		46.509	46.709	46.795	46.771	43.165	43.465	43.307	50.421
SiH ₄	8.20		10.934	11.302	11.422	11.436	5.678	6.161	8.051	16.150
² PH ₂	33.10		27.532	29.353	29.389	29.341	26.007	27.951	30.299	31.274
PH ₃	1.30		-0.138	2.117	2.198	2.092	-2.698	-0.249	2.916	3.543
H ₂ S	-4.90		-4.547	-3.322	-3.314	-3.590	-3.730	-2.315	-4.802	-4.615
HCl	-22.00		-19.767	-20.186	-20.189	-20.376	-18.195	-18.496	-21.697	-21.934
Li ₂	51.60		55.210	55.257	55.253	55.260	57.583	57.626	57.947	56.191
LiF	-80.10		-71.221	-74.047	-73.943	-73.997	-68.016	-70.754	-75.223	-79.231
HC≡CH	54.60		56.056	54.952	54.842	54.378	57.908	56.714	57.291	45.137
H ₂ C=CH ₂	12.50		12.473	14.163	14.063	13.355	12.510	14.350	13.271	4.228
H ₃ C-CH ₃	-20.10		-18.685	-15.776	-15.872	-16.808	-18.341	-15.200	-19.899	-25.214
CN	105.20		97.117	96.064	95.991	95.935	99.331	98.190	108.285	89.042
HCN	30.90		26.818	26.609	26.513	26.299	27.352	27.082	36.347	18.183
CO	-26.40		-24.030	-26.272	-26.197	-26.385	-22.623	-24.768	-22.494	-36.038
² HCO	10.00		4.887	3.481	3.544	3.252	7.436	6.123	6.377	-6.433
H ₂ CO	-26.10		-25.914	-26.645	-26.585	-26.994	-25.283	-25.910	-25.490	-38.135
CH ₃ OH	-48.00		-44.995	-44.358	-44.273	-44.970	-42.189	-41.343	-46.549	-55.038
N ₂	0.00		-9.460	-8.139	-8.310	-8.285	-10.606	-9.381	8.712	-14.907
H ₂ NNH ₂	23.30		17.194	21.697	21.633	21.013	20.193	24.896	32.413	12.788
NO	21.80		10.387	11.001	10.969	10.953	11.044	11.676	22.182	2.200
³ O ₂	0.00		-11.362	-12.034	-11.978	-12.129	-8.723	-9.316	-9.495	-23.578
H ₂ O ₂	-32.40		-34.134	-33.284	-33.107	-33.590	-29.115	-27.951	-31.281	-45.264
F ₂	0.00		-4.972	-6.102	-5.876	-5.741	-5.078	-5.948	-0.542	-14.480
CO ₂	-94.00		-96.615	-101.785	-101.750	-102.179	-92.242	-97.391	-99.730	-121.233
Na ₂	34.00		33.438	33.542	33.547	33.550	36.049	36.162	37.360	33.232
Si ₂	139.90		139.133	137.328	137.319	137.345	141.270	139.427	140.050	133.292
P ₂	34.30		31.796	31.593	31.588	31.739	31.492	31.340	38.906	29.927
³ S ₂	30.70		22.098	21.222	21.211	21.184	24.059	23.224	21.557	17.177

(continued)

Cl ₂	0.00	-0.869	-2.089	-1.991	-1.912	-0.865	-1.919	-1.173	-7.570
NaCl	-43.60	-38.153	-40.030	-39.969	-40.077	-36.560	-38.386	-42.209	-39.225
SiO	-24.60	-19.203	-21.605	-21.404	-21.479	-17.387	-19.597	-19.627	-27.933
CS	66.90	67.008	65.138	65.091	64.991	68.130	66.335	69.398	58.405
SO	1.20	-5.856	-6.848	-6.696	-6.800	-3.628	-4.428	-5.788	-14.255
ClO	24.30	15.738	15.584	15.690	15.703	16.652	16.666	18.816	7.982
ClF	-13.30	-14.689	-16.160	-15.961	-15.846	-14.769	-15.989	-13.513	-23.623
Si ₂ H ₆	19.10	20.408	21.827	22.010	21.856	11.356	13.019	16.115	28.424
CH ₃ Cl	-19.60	-18.575	-18.502	-18.498	-18.921	-18.492	-18.311	-20.538	-25.231
H ₃ C-SH	-5.50	-5.523	-3.735	-3.774	-4.374	-5.148	-3.086	-6.357	-10.501
HOCl	-18.40	-19.367	-19.592	-19.445	-19.646	-16.849	-16.816	-18.260	-28.359
SO ₂	-71.00	-71.260	-74.905	-74.613	-74.828	-68.025	-71.332	-72.660	-91.183
BF ₃	-271.40	-258.398	-267.343	-266.884	-267.240	-253.001	-261.526	-273.732	-282.638
BCl ₃	-96.30	-96.457	-101.249	-101.186	-101.635	-93.220	-97.696	-106.297	-110.171
AlF ₃	-289.00	-269.775	-278.657	-278.277	-278.523	-265.504	-274.062	-286.282	-287.468
AlCl ₃	-139.70	-135.310	-141.287	-141.054	-141.410	-132.864	-138.614	-147.936	-140.648
CF ₄	-223.10	-217.478	-228.411	-227.853	-227.998	-212.131	-222.543	-229.516	-250.932
CCl ₄	-22.90	-25.064	-30.435	-30.434	-30.539	-22.652	-27.643	-29.324	-44.635
OCS	-33.10	-40.292	-44.547	-44.541	-44.882	-36.797	-40.958	-42.142	-59.555
CS ₂	28.00	18.381	14.678	14.598	14.337	21.373	17.758	16.922	3.687
F ₂ CO	-145.00	-144.300	-152.159	-151.854	-152.134	-139.523	-147.085	-151.353	-174.364
SiF ₄	-386.00	-361.077	-373.161	-372.539	-372.791	-356.152	-367.703	-381.350	-383.670
SiCl ₄	-158.40	-152.616	-160.212	-159.900	-160.226	-150.085	-157.307	-165.540	-162.572
NNO	19.70	-1.347	-2.910	-3.102	-3.161	0.400	-1.311	14.882	-20.819
ClNO	12.60	-5.187	-6.938	-6.901	-6.898	-3.620	-5.256	5.366	-21.066
NF ₃	-31.60	-49.226	-53.398	-53.044	-52.785	-47.882	-51.654	-40.578	-71.956
PF ₃	-229.10	-220.179	-227.436	-226.830	-226.748	-218.215	-224.894	-226.170	-239.427
O ₃	33.90	19.396	18.388	18.573	18.537	21.805	21.096	29.247	-3.619
F ₂ O	5.90	-8.284	-9.714	-9.364	-9.209	-7.902	-8.912	-1.016	-24.698
ClF ₃	-38.00	-51.706	-57.418	-56.919	-56.625	-48.937	-54.115	-52.045	-76.974
F ₂ C=CF ₂	-161.30	-162.865	-173.293	-172.793	-173.119	-157.518	-167.401	-170.535	-204.597
Cl ₂ C=CCl ₂	-5.50	-10.301	-16.504	-16.672	-17.032	-7.064	-12.968	-14.830	-38.126
F ₃ C-CN	-118.40	-124.320	-132.204	-131.919	-132.286	-120.187	-127.682	-121.756	-160.508
HC≡C-CH ₃	44.40	44.298	44.127	43.923	42.999	46.741	46.624	44.560	28.038
H ₂ C=C=CH ₂	45.40	40.595	41.211	40.997	40.054	42.603	43.372	41.757	24.612
C ₃ H ₄ (cyclopropene)	67.80	63.614	63.715	63.498	62.712	65.251	65.508	65.517	47.295
H ₂ C=CH-CH ₃	4.80	3.765	6.666	6.483	5.334	4.236	7.432	4.184	-8.811
C ₃ H ₆ (cyclopropane)	12.80	10.652	12.384	12.159	11.141	11.652	13.638	10.333	-2.957
CH ₃ -CH ₂ -CH ₃	-25.10	-24.064	-19.883	-20.048	-21.389	-23.756	-19.191	-25.256	-34.289
C ₄ H ₆ (Z-1,3-butadiene)	26.40	22.152	25.167	24.901	23.547	22.389	25.744	24.253	3.800
C ₄ H ₆ (2-butyne)	34.80	34.304	34.998	34.699	33.334	37.176	38.067	33.601	12.610
C ₄ H ₆ (methylenecyclopropane)	47.90	40.161	41.648	41.325	40.080	41.972	43.782	41.192	19.512
C ₄ H ₆ (bicyclo[1.1.0]butane)	51.90	46.936	47.503	47.171	46.081	48.920	49.685	48.326	26.702
C ₄ H ₆ (cyclobutene)	38.20	33.235	36.481	36.265	34.936	33.373	37.117	36.791	14.490
C ₄ H ₈ (cyclobutane)	6.60	3.500	8.128	7.924	6.395	3.135	8.349	5.204	-12.675
C ₄ H ₈ (isobutene)	-4.10	-5.710	-1.527	-1.784	-3.363	-4.892	-0.281	-5.292	-21.712
C ₄ H ₁₀ (trans butane)	-30.10	-29.393	-23.955	-24.188	-25.935	-29.122	-23.148	-30.572	-43.284
C ₄ H ₁₀ (isobutane)	-32.20	-31.423	-25.913	-26.139	-27.880	-31.056	-25.019	-32.187	-44.374
C ₅ H ₈ (spiropentane)	44.30	37.107	39.006	38.575	37.049	39.162	41.479	37.527	11.935
C ₆ H ₆ (benzene)	19.90	7.780	11.253	10.751	8.821	8.284	12.330	10.403	-22.848
CH ₂ F ₂	-107.70	-104.204	-108.414	-108.092	-108.337	-103.184	-107.109	-109.028	-122.140
CHF ₃	-166.30	-162.018	-169.598	-169.145	-169.322	-159.077	-166.249	-170.262	-187.903

(continued)

CH ₂ Cl ₂	-22.60	-22.346	-24.127	-24.119	-24.440	-21.910	-23.521	-25.527	-33.914
CHCl ₃	-24.50	-25.176	-28.817	-28.816	-29.034	-23.854	-27.233	-29.246	-41.183
CH ₃ NH ₂	-5.20	-0.375	3.134	3.078	2.289	0.917	4.643	5.683	-5.273
CH ₃ CN	18.00	11.856	12.545	12.359	11.678	12.990	13.763	20.217	-2.068
CH ₃ NO ₂ (nitromethane)	-17.90	-35.174	-35.093	-35.135	-35.724	-33.259	-32.960	-25.879	-59.555
CH ₃ ONO (methyl nitrite)	-16.10	-32.835	-32.874	-32.778	-33.238	-31.509	-31.293	-21.287	-54.847
CH ₃ SiH ₃	-7.00	-3.502	-1.984	-1.883	-2.416	-6.960	-5.122	-7.007	-2.305
HCO ₂ H	-90.40	-90.503	-93.200	-93.062	-93.710	-85.567	-88.011	-93.046	-111.954
HCO ₂ CH ₃	-85.50	-88.070	-89.998	-89.807	-90.647	-85.711	-87.341	-89.540	-111.885
CH ₃ CONH ₂	-57.00	-64.185	-62.589	-62.664	-63.867	-59.880	-57.904	-58.876	-84.231
C ₂ H ₅ N (aziridine)	30.20	21.742	24.060	23.947	23.195	22.658	25.197	30.296	10.031
NCCN (cyanogen)	74.10	57.991	56.920	56.628	56.248	59.282	58.096	77.596	37.081
NH(CH ₃) ₂	-4.30	-8.323	-4.227	-4.316	-5.366	-7.916	-3.513	-1.721	-17.806
CH ₃ CH ₂ NH ₂	-11.30	-15.894	-11.270	-11.398	-12.582	-14.069	-9.052	-9.440	-25.203
H ₂ C=C=O (ketene)	-11.60	-16.890	-19.237	-19.314	-20.013	-13.082	-15.299	-17.674	-37.267
C ₂ H ₄ O (oxirane)	-12.60	-16.956	-16.660	-16.563	-17.208	-16.229	-15.658	-15.115	-32.026
CH ₃ CHO	-39.50	-40.873	-40.396	-40.420	-41.303	-39.372	-38.611	-41.209	-57.475
O=CH-CH=O	-50.80	-54.597	-56.230	-56.157	-56.944	-52.833	-54.157	-52.935	-80.457
CH ₃ CH ₂ OH	-56.10	-53.946	-51.946	-51.927	-53.052	-50.992	-48.616	-55.501	-67.563
(CH ₃) ₂ O	-44.00	-43.489	-42.119	-41.997	-42.887	-43.151	-41.531	-44.134	-56.457
C ₂ H ₄ S (thiooxirane)	19.60	14.486	15.219	15.142	14.512	14.950	15.899	15.131	2.869
(CH ₃) ₂ S=O	-36.20	-38.239	-36.977	-36.976	-38.054	-36.313	-34.504	-40.593	-55.337
CH ₃ CH ₂ SH	-11.10	-11.299	-8.284	-8.389	-9.389	-10.921	-7.478	-12.118	-19.965
(CH ₃) ₂ S	-8.90	-9.578	-7.235	-7.328	-8.260	-9.362	-6.644	-10.926	-19.041
H ₂ C=CHF	-34.00	-34.775	-36.014	-35.950	-36.561	-33.487	-34.467	-35.880	-51.223
CH ₃ CH ₂ Cl	-26.60	-26.379	-25.055	-25.114	-25.953	-26.168	-24.574	-28.392	-36.728
H ₂ C=CHCl	5.20	3.547	3.291	3.183	2.545	4.367	4.300	2.760	-9.818
H ₂ C=CHCN	43.20	35.738	36.547	36.280	35.411	36.501	37.435	46.042	15.936
(CH ₃) ₂ C=O	-51.70	-53.966	-52.199	-52.295	-53.634	-51.799	-49.578	-54.629	-73.961
CH ₃ CO ₂ H	-103.40	-104.124	-105.368	-105.301	-106.424	-98.510	-99.338	-106.948	-128.749
CH ₃ CFO	-105.70	-106.396	-109.301	-109.179	-109.984	-103.017	-105.555	-109.997	-131.162
CH ₃ COCl	-57.70	-61.889	-63.982	-63.993	-64.800	-58.797	-60.610	-64.631	-83.546
CH ₃ CH ₂ CH ₂ Cl	-31.50	-31.636	-29.087	-29.214	-30.455	-31.474	-28.506	-33.686	-45.698
(CH ₃) ₂ CHOH	-65.20	-63.969	-60.593	-60.636	-62.180	-60.734	-56.820	-65.178	-80.323
CH ₃ -O-CH ₂ CH ₃	-51.70	-52.283	-49.553	-49.498	-50.818	-51.787	-48.641	-52.970	-68.837
(CH ₃) ₃ N	-6.60	-11.277	-6.450	-6.558	-7.875	-12.139	-6.923	-4.148	-23.659
C ₄ H ₄ O (furan)	-8.30	-17.416	-17.826	-17.894	-19.124	-15.088	-15.022	-15.438	-46.264
C ₄ H ₄ S (thiophene)	27.50	18.128	18.525	18.184	16.937	19.832	20.730	18.924	-6.991
C ₄ H ₅ N (pyrrole)	25.90	11.965	14.081	13.806	12.408	14.275	16.834	20.536	-13.692
C ₅ H ₅ N (pyridine)	33.60	15.585	18.805	18.374	16.784	15.252	19.005	25.893	-14.365
H ₂	0.00	1.322	1.322	1.322	1.322	3.877	3.877	2.029	4.886
² SH	34.20	32.088	33.096	33.094	32.954	32.629	33.713	32.445	33.105
² C≡CH	135.80	134.024	132.370	132.258	131.913	136.932	135.206	134.490	123.889
HC=CH ₂ (² A')	71.00	65.965	66.756	66.638	66.034	67.630	68.485	66.139	58.415
CH ₃ C=O (² A')	-2.40	-8.747	-8.994	-8.999	-9.746	-5.786	-5.808	-8.016	-24.710
CH ₂ -OH (² A)	-4.00	-5.907	-6.037	-5.957	-6.549	-2.063	-2.017	-7.899	-15.885
CH ₃ O (² A')	5.20	-1.823	-0.487	-0.441	-0.959	-0.553	0.953	-1.003	-8.513
CH ₃ CH ₂ O (² A'')	-2.90	-11.455	-8.857	-8.881	-9.834	-9.810	-6.862	-10.758	-22.343
CH ₃ S (² A')	29.80	25.697	27.136	27.084	26.605	26.252	27.874	25.434	21.234
CH ₂ CH ₃ (² A')	28.60	26.062	28.013	27.898	27.057	27.160	29.235	23.478	19.913
(CH ₃) ₂ CH (² A')	21.10	17.254	20.434	20.243	18.963	18.827	22.260	14.611	6.724
² C(CH ₃) ₃	12.00	7.741	12.181	11.922	10.214	9.683	14.501	5.238	-6.483

(continued)

NO ₂ (² A ₁)	8.10	-13.683	-15.203	-15.256	-15.378	-10.710	-12.234	-3.219	-34.634
CH ₃ CH=C=CH ₂	38.80	33.842	35.560	35.268	33.913	35.948	37.954	34.599	13.314
C ₅ H ₈ (isoprene)	18.00	13.269	17.580	17.244	15.460	13.787	18.578	15.537	-8.185
C ₅ H ₁₀ (cyclopentane twist)	-18.50	-20.124	-14.267	-14.579	-16.576	-20.360	-13.738	-20.086	-40.154
C ₅ H ₁₂ (n-pentane)	-35.00	-34.655	-27.959	-28.262	-30.413	-34.432	-27.050	-35.811	-52.206
C(CH ₃) ₄ (neopentane)	-40.10	-39.945	-33.063	-33.340	-35.469	-39.519	-31.976	-39.783	-54.531
C ₆ H ₈ (1,3-cyclohexadiene)	25.40	19.087	23.832	23.377	21.287	19.648	25.105	21.575	-9.671
C ₆ H ₈ (1,4-cyclohexadiene)	25.00	19.657	24.359	23.891	21.782	20.442	25.863	21.779	-9.517
C ₆ H ₁₂ (cyclohexane chair)	-29.40	-30.939	-23.522	-23.923	-26.350	-31.172	-22.844	-30.893	-53.494
C ₆ H ₁₄ (n-hexane)	-39.90	-39.936	-31.983	-32.355	-34.910	-39.757	-30.966	-41.090	-61.144
C ₆ H ₁₄ (3-methylpentane)	-41.10	-40.484	-32.493	-32.852	-35.399	-40.342	-31.512	-41.004	-60.442
C ₆ H ₅ -CH ₃ (toluene)	12.00	-0.311	4.379	3.800	1.436	0.630	6.023	2.192	-34.726
C ₇ H ₁₆ (n-heptane)	-44.80	-45.208	-35.999	-36.439	-39.399	-45.074	-34.875	-46.317	-70.074
C ₈ H ₈ (cyclooctatetraene)	70.70	59.547	64.454	63.800	61.253	60.564	66.287	65.076	18.532
C ₈ H ₁₈ (n-octane)	-49.80	-50.491	-40.024	-40.534	-43.898	-50.401	-38.794	-51.578	-79.015
C ₁₀ H ₈ (naphthalene)	35.90	12.133	17.592	16.698	13.551	13.144	19.578	17.424	-40.015
C ₁₀ H ₈ (azulene)	69.10	43.747	49.017	48.147	45.047	45.091	51.337	50.165	-9.667
CH ₃ CO ₂ CH ₃ (Z-methylacetate)	-98.40	-101.175	-101.667	-101.548	-102.858	-98.182	-98.209	-102.893	-128.162
(CH ₃) ₃ COH (t-butanol)	-74.70	-74.214	-69.403	-69.498	-71.452	-70.725	-65.227	-74.627	-92.259
C ₆ H ₅ NH ₂ (aniline)	20.80	1.796	6.592	6.037	3.807	4.965	10.447	11.151	-32.284
C ₆ H ₅ OH (phenol)	-22.30	-34.941	-32.859	-33.266	-35.426	-30.508	-27.717	-33.497	-73.497
C ₄ H ₆ O (divinyl ether)	-3.30	-8.750	-7.904	-7.958	-9.333	-6.590	-5.319	-7.406	-35.231
C ₄ H ₈ O (tetrahydrofuran)	-44.00	-46.020	-42.825	-42.859	-44.443	-45.871	-42.019	-45.741	-68.725
C ₅ H ₈ O (cyclopentanone)	-45.90	-51.375	-48.082	-48.342	-50.348	-49.466	-45.342	-51.377	-81.653
C ₆ H ₄ O ₂ (benzoquinone)	-29.40	-43.777	-42.927	-43.227	-45.260	-41.541	-39.911	-38.774	-90.551
C ₆ H ₄ N ₂ (pyrimidine)	46.80	21.576	24.416	24.048	22.800	20.482	23.810	39.251	-7.724
(CH ₃) ₂ SO ₂	-89.20	-89.810	-90.004	-89.919	-91.247	-85.931	-85.520	-95.574	-113.367
C ₆ H ₅ Cl (chlorobenzene)	12.50	-1.026	0.454	-0.051	-1.900	0.291	2.376	0.253	-36.164
NC(CH ₂) ₂ CN (succinonitrile)	50.10	36.718	37.708	37.301	36.084	38.262	39.480	54.670	7.507
C ₄ H ₄ N ₂ (pyrazine)	46.90	25.091	28.171	27.817	26.567	23.921	27.497	43.407	-4.086
C ₄ H ₄ O (3-buten-2-one)	15.60	13.813	12.712	12.512	11.223	17.502	16.643	15.244	-15.729
C ₄ H ₆ O (E-crotonaldehyde)	-24.00	-30.522	-28.702	-28.889	-30.407	-29.057	-26.767	-29.433	-57.070
C ₄ H ₆ O ₃ (acetic anhydride)	-136.80	-143.885	-146.238	-146.158	-147.889	-137.906	-139.608	-146.708	-185.596
C ₄ H ₆ S (2,5-dihydrothiophene)	20.80	14.933	17.559	17.289	15.868	15.207	18.484	15.945	-7.106
(CH ₃) ₂ CHCN	5.60	0.675	3.919	3.613	2.144	1.574	5.187	9.229	-20.146
C ₄ H ₈ O (methyl ethyl ketone)	-57.10	-59.733	-56.712	-56.875	-58.619	-57.576	-53.944	-60.325	-83.160
(CH ₃) ₂ CHCHO	-51.60	-52.191	-49.283	-49.434	-51.098	-50.804	-47.331	-52.368	-75.639
C ₄ H ₈ O ₂ (1,4-dioxane)	-75.50	-79.758	-77.661	-77.480	-79.077	-79.218	-76.407	-78.669	-107.583
C ₄ H ₈ S (tetrahydrothiophene)	-8.20	-10.845	-6.875	-7.109	-8.689	-11.100	-6.394	-10.968	-29.721
(CH ₃) ₃ CCl	-43.50	-45.017	-40.904	-41.065	-42.717	-44.447	-39.763	-45.684	-59.576
C ₄ H ₉ Cl (n-butyl chloride)	-37.00	-36.835	-33.033	-33.228	-34.871	-36.740	-32.370	-38.869	-54.560
C ₄ H ₉ N (tetrahydropyrrole)	-0.80	-8.067	-2.167	-2.393	-4.117	-8.140	-1.532	-0.164	-27.032
C ₄ H ₉ NO ₂ (2-nitrobutane)	-39.10	-57.293	-53.427	-53.657	-55.462	-55.156	-50.604	-47.460	-90.951
(CH ₃ CH ₂) ₂ O	-60.30	-60.959	-56.874	-56.886	-58.634	-60.320	-55.651	-61.704	-81.093
CH ₃ CH(OCH ₃) ₂ (dimethyl acetal)	-93.10	-94.781	-92.514	-92.304	-94.010	-93.481	-90.591	-95.091	-119.944
(CH ₃) ₃ CSH	-26.20	-28.123	-22.438	-22.649	-24.433	-27.482	-21.083	-27.472	-41.107
C ₄ H ₁₀ S ₂ (diethyl disulfide)	-17.90	-20.993	-16.792	-17.002	-18.709	-20.505	-15.554	-22.447	-41.111
(CH ₃) ₃ CNH ₂	-28.90	-34.252	-26.808	-27.045	-29.039	-32.125	-24.000	-26.378	-47.647
(CH ₃) ₄ Si	-55.70	-49.712	-44.601	-44.539	-46.710	-47.136	-41.109	-54.175	-58.530
C ₅ H ₆ S (2-methyl thiophene)	20.00	10.233	11.809	11.390	9.717	12.277	14.483	10.745	-18.880
C ₅ H ₇ N (N-methyl pyrrole)	24.60	10.345	13.132	12.838	11.152	11.465	14.779	18.955	-19.073
C ₅ H ₁₀ O (tetrahydropyran)	-53.40	-56.001	-51.304	-51.416	-53.428	-55.810	-50.295	-55.557	-81.279

(continued)

$(\text{CH}_3\text{CH}_2)_2\text{C}=\text{O}$	-61.60	-65.396	-61.130	-61.358	-63.507	-63.262	-58.227	-65.921	-92.252
$\text{C}_5\text{H}_{10}\text{O}_2$ (isopropyl acetate)	-115.10	-119.100	-116.863	-116.873	-119.037	-115.704	-112.678	-120.534	-152.246
$\text{C}_5\text{H}_{10}\text{S}$ (tetrahydrothiopyran)	-15.20	-18.290	-12.763	-13.089	-15.104	-18.488	-12.055	-18.399	-40.167
$\text{C}_5\text{H}_{11}\text{N}$ (piperidine)	-11.30	-18.181	-10.877	-11.196	-13.345	-18.020	-9.852	-10.425	-39.964
$\text{C}_5\text{H}_{12}\text{O}$ (t-butyl methyl ether)	-67.80	-69.190	-63.659	-63.713	-65.858	-68.289	-62.025	-68.525	-89.665
$\text{C}_6\text{H}_4\text{F}_2$ (1,3-difluorobenzene)	-73.90	-87.740	-90.078	-90.254	-92.018	-84.300	-85.869	-88.648	-133.695
$\text{C}_6\text{H}_4\text{F}_2$ (1,4-difluorobenzene)	-73.30	-86.982	-89.302	-89.478	-91.237	-83.608	-85.164	-87.757	-132.983
$\text{C}_6\text{H}_5\text{F}$ (fluorobenzene)	-27.40	-40.311	-39.750	-40.088	-41.933	-38.346	-37.115	-39.430	-78.603
$\text{C}_6\text{H}_{14}\text{O}$ (diisopropyl ether)	-76.30	-78.020	-71.166	-71.295	-73.875	-77.025	-69.263	-77.839	-103.079
PF_5	-381.10	-364.825	-377.789	-376.790	-376.879	-359.910	-371.940	-380.254	-391.911
SF_6	-291.70	-289.541	-303.559	-302.412	-302.338	-283.071	-295.991	-303.621	-323.703
P_4	14.10	-4.250	-4.575	-4.554	-4.063	-4.603	-4.811	9.944	-6.429
SO_3	-94.60	-97.506	-103.477	-103.157	-103.549	-91.963	-97.609	-102.558	-125.370
SCl_2	-4.20	-7.894	-10.166	-10.036	-9.944	-7.254	-9.299	-9.243	-18.677
POCl_3	-133.80	-134.339	-140.596	-140.310	-140.495	-131.249	-137.130	-140.862	-151.623
PCl_5	-86.10	-94.488	-101.034	-100.744	-100.696	-91.907	-98.031	-99.199	-108.417
$\text{Cl}_2\text{O}_2\text{S}$	-84.80	-89.446	-96.189	-95.918	-96.150	-84.102	-90.533	-96.999	-116.410
PCl_3	-69.00	-72.323	-76.457	-76.270	-76.188	-70.937	-74.824	-74.777	-83.704
Cl_2S_2	-4.00	-15.860	-19.236	-19.103	-19.007	-13.790	-16.868	-18.120	-31.124
SiCl_2 (1A_1)	-40.30	-37.993	-42.201	-42.012	-42.066	-36.377	-40.401	-42.734	-46.440
CF_3Cl	-169.70	-167.901	-177.302	-176.882	-177.026	-163.341	-172.236	-177.683	-197.941
C_2F_6	-320.90	-318.228	-333.038	-332.161	-332.512	-311.386	-325.326	-332.636	-367.896
$^2\text{CF}_3$	-111.80	-113.522	-121.704	-121.275	-121.388	-109.001	-116.788	-122.127	-139.910
$^2\text{C}_6\text{H}_5$ (phenyl radical)	80.50	63.114	65.567	65.060	63.260	65.129	68.046	65.334	33.117
ME		-3.790	-3.467	-3.512	-4.411	-2.208	-1.520	-3.145	-20.878
MAD		5.612	5.885	5.895	6.128	5.452	5.354	4.488	21.385

TABLE X: Bond lengths in Å for the T96-R molecular test set [8, 9] from different approximate XC functionals.

Molecule	Exp.	X	<i>rreg</i> TM	<i>v1-sreg</i> TM	<i>v2-sreg</i> TM	<i>v3-sreg</i> TM	<i>rreg</i> TM	<i>v2-sreg</i> TM	r ² SCAN	PBE
		C	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	PBE	PBE	r ² SCAN	PBE
H ₂	0.741		0.744	0.744	0.744	0.744	0.743	0.743	0.742	0.751
Li ₂	2.673		2.731	2.736	2.736	2.736	2.738	2.744	2.762	2.738
LiH	1.595		1.606	1.606	1.606	1.607	1.595	1.595	1.605	1.607
LiF	1.564		1.579	1.579	1.579	1.579	1.571	1.571	1.574	1.576
LiCl	2.021		2.029	2.029	2.029	2.029	2.018	2.018	2.029	2.021
LiO	1.688		1.702	1.703	1.703	1.702	1.691	1.692	1.696	1.699
Be ₂	2.440		2.486	2.487	2.487	2.486	2.439	2.428	2.489	2.441
BeH	1.343		1.359	1.360	1.360	1.360	1.351	1.351	1.357	1.362
BeF	1.361		1.381	1.381	1.381	1.380	1.378	1.378	1.374	1.382
BeO	1.331		1.344	1.344	1.344	1.344	1.340	1.340	1.335	1.346
BeS	1.742		1.754	1.754	1.754	1.754	1.748	1.749	1.749	1.759
B ₂	1.590		1.622	1.621	1.621	1.620	1.618	1.617	1.618	1.619
BH	1.232		1.245	1.245	1.245	1.245	1.237	1.237	1.242	1.252
BF	1.263		1.274	1.273	1.273	1.273	1.272	1.272	1.267	1.276
BF ₃	1.313		1.322	1.321	1.321	1.321	1.320	1.319	1.314	1.325
BCl	1.715		1.725	1.729	1.728	1.728	1.719	1.722	1.724	1.732
BCl ₃	1.742		1.743	1.744	1.744	1.744	1.741	1.742	1.742	1.748
BN	1.281		1.329	1.331	1.331	1.331	1.324	1.326	1.321	1.333
BO	1.204		1.213	1.213	1.213	1.213	1.210	1.210	1.206	1.215
BS	1.609		1.617	1.618	1.618	1.617	1.613	1.614	1.612	1.621
C ₂	1.242		1.254	1.399	1.399	1.254	1.252	1.253	1.249	1.404
CH	1.120		1.135	1.136	1.136	1.136	1.126	1.127	1.126	1.137
CH ₄	1.087		1.092	1.093	1.093	1.092	1.089	1.090	1.088	1.096
CF	1.272		1.288	1.287	1.287	1.286	1.286	1.285	1.278	1.287
CF ₄	1.323		1.334	1.332	1.332	1.332	1.333	1.331	1.325	1.337
CCl	1.645		1.663	1.663	1.663	1.662	1.661	1.661	1.654	1.660
CCl ₄	1.767		1.777	1.778	1.778	1.778	1.773	1.775	1.774	1.782
CN	1.172		1.172	1.173	1.173	1.173	1.170	1.171	1.165	1.174
CO	1.128		1.133	1.134	1.134	1.134	1.132	1.133	1.127	1.136
CO ⁺	1.115		1.120	1.120	1.120	1.120	1.117	1.118	1.112	1.122
CO ₂	1.160		1.169	1.169	1.169	1.169	1.168	1.168	1.162	1.171
CP	1.562		1.562	1.562	1.562	1.562	1.558	1.559	1.555	1.567
CS	1.535		1.541	1.543	1.543	1.542	1.538	1.540	1.536	1.546
CS ₂	1.553		1.559	1.559	1.559	1.558	1.557	1.557	1.553	1.562
N ₂	1.098		1.099	1.101	1.101	1.100	1.098	1.099	1.093	1.103
N ₂ ⁺	1.116		1.113	1.114	1.113	1.113	1.111	1.112	1.106	1.114
NH	1.036		1.049	1.049	1.049	1.049	1.043	1.043	1.041	1.051
NH ⁺	1.070		1.087	1.088	1.088	1.088	1.078	1.080	1.078	1.091
NF	1.317		1.332	1.329	1.329	1.328	1.331	1.328	1.319	1.327
NCl	1.611		1.626	1.624	1.623	1.622	1.625	1.622	1.616	1.618
NO	1.151		1.157	1.157	1.157	1.156	1.155	1.155	1.148	1.158
NO ⁺	1.063		1.066	1.067	1.067	1.067	1.064	1.065	1.059	1.070
NS	1.494		1.504	1.504	1.504	1.503	1.502	1.501	1.494	1.506
O ₂	1.208		1.220	1.219	1.218	1.218	1.219	1.218	1.207	1.219
O ₂ ⁺	1.116		1.121	1.120	1.120	1.120	1.119	1.119	1.109	1.122
OH	0.970		0.981	0.982	0.982	0.981	0.978	0.979	0.973	0.983
OH ⁺	1.029		1.044	1.044	1.044	1.044	1.037	1.037	1.035	1.048
OF	1.358		1.367	1.364	1.364	1.364	1.368	1.365	1.350	1.360
F ₂	1.412		1.417	1.415	1.415	1.414	1.417	1.415	1.399	1.413

(continued)

F_2^+	1.322	1.317	1.315	1.315	1.315	1.317	1.315	1.298	1.316
HF	0.917	0.929	0.929	0.929	0.928	0.927	0.927	0.921	0.930
HF^+	1.001	1.019	1.020	1.020	1.019	1.014	1.015	1.010	1.023
Na_2	3.079	3.079	3.079	3.079	3.079	3.131	3.130	3.142	3.079
NaH	1.887	1.891	1.892	1.892	1.892	1.887	1.887	1.898	1.893
NaF	1.926	1.931	1.931	1.931	1.930	1.929	1.929	1.926	1.940
NaCl	2.361	2.361	2.361	2.361	2.361	2.358	2.358	2.365	2.367
NaO	2.052	2.059	2.059	2.059	2.059	2.055	2.056	2.056	2.070
MgH	1.730	1.750	1.751	1.751	1.751	1.735	1.736	1.745	1.757
MgF	1.750	1.775	1.774	1.774	1.773	1.771	1.771	1.766	1.784
MgCl	2.196	2.215	2.214	2.214	2.214	2.209	2.209	2.212	2.226
MgO	1.748	1.748	1.744	1.744	1.744	1.747	1.743	1.736	1.750
Al_2	2.466	2.473	2.470	2.470	2.469	2.470	2.466	2.471	2.488
AlH	1.648	1.666	1.666	1.666	1.666	1.654	1.654	1.661	1.679
AlF	1.654	1.675	1.674	1.674	1.673	1.672	1.671	1.664	1.684
AlCl	2.130	2.150	2.149	2.149	2.149	2.143	2.143	2.145	2.163
AlO	1.618	1.633	1.629	1.629	1.629	1.631	1.628	1.619	1.633
AlS	2.029	2.042	2.040	2.040	2.039	2.038	2.035	2.031	2.049
Si_2	2.246	2.155	2.157	2.157	2.156	2.149	2.152	2.152	2.287
SiH	1.520	1.538	1.537	1.538	1.538	1.527	1.526	1.530	1.545
SiH_4	1.480	1.486	1.485	1.485	1.485	1.481	1.481	1.482	1.493
SiF	1.601	1.625	1.623	1.623	1.623	1.622	1.620	1.613	1.632
SiF_4	1.553	1.574	1.572	1.572	1.572	1.572	1.571	1.564	1.580
SiCl	2.058	2.077	2.076	2.076	2.075	2.074	2.072	2.070	2.084
$SiCl_4$	2.019	2.029	2.028	2.028	2.028	2.026	2.026	2.025	2.039
SiN	1.572	1.576	1.575	1.575	1.575	1.573	1.572	1.569	1.580
SiO	1.510	1.524	1.524	1.524	1.523	1.522	1.521	1.515	1.529
SiS	1.929	1.942	1.942	1.942	1.942	1.937	1.938	1.937	1.951
P_2	1.893	1.896	1.897	1.897	1.897	1.891	1.893	1.891	1.905
P_4	2.210	2.199	2.199	2.198	2.198	2.195	2.194	2.190	2.208
PH	1.421	1.436	1.435	1.436	1.435	1.427	1.427	1.427	1.440
PF	1.589	1.611	1.609	1.609	1.608	1.609	1.607	1.599	1.617
PCl	2.015	2.028	2.026	2.025	2.025	2.026	2.023	2.019	2.031
PN	1.491	1.495	1.496	1.496	1.495	1.492	1.493	1.487	1.500
PO	1.476	1.493	1.492	1.492	1.491	1.490	1.489	1.482	1.497
S_2	1.889	1.906	1.905	1.905	1.904	1.904	1.902	1.896	1.912
SH	1.341	1.351	1.352	1.352	1.352	1.345	1.346	1.345	1.356
SF	1.601	1.621	1.618	1.618	1.617	1.620	1.616	1.605	1.622
SF_6	1.561	1.589	1.585	1.586	1.585	1.588	1.584	1.576	1.596
SO	1.481	1.503	1.501	1.501	1.501	1.501	1.499	1.490	1.505
SO_3	1.420	1.441	1.439	1.439	1.438	1.439	1.437	1.429	1.444
Cl_2	1.988	2.013	2.009	2.009	2.008	2.012	2.008	1.996	2.008
Cl_2^+	1.891	1.915	1.911	1.911	1.910	1.913	1.910	1.898	1.914
HCl	1.275	1.283	1.285	1.285	1.285	1.279	1.281	1.279	1.288
HCl^+	1.315	1.327	1.329	1.329	1.329	1.320	1.322	1.321	1.333
ClF	1.628	1.655	1.651	1.651	1.651	1.654	1.650	1.636	1.652
ClO	1.570	1.594	1.589	1.589	1.588	1.595	1.589	1.575	1.583
ME		0.012	0.013	0.013	0.011	0.008	0.008	0.005	0.018
MAD		0.014	0.015	0.015	0.014	0.011	0.011	0.010	0.018

TABLE XI: Harmonic vibrational frequencies in cm^{-1} for the T82-F molecular test set [8, 9] obtained with different approximate exchange-correlation functionals.

Molecule	Exp.	X	<i>rreg</i> TM	<i>v1-sreg</i> TM	<i>v2-sreg</i> TM	<i>v3-sreg</i> TM	<i>rreg</i> TM	<i>v2-sreg</i> TM	r ² SCAN	PBE
		C	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	<i>rreg</i> TM	PBE	PBE	r ² SCAN	PBE
H ₂	4401.20		4387.64	4387.61	4387.61	4387.55	4423.31	4423.28	4429.67	4316.60
Li ₂	351.40		328.66	329.72	329.64	329.92	338.35	338.47	332.37	329.83
LiH	1405.70		1384.23	1383.65	1383.55	1383.36	1411.41	1411.80	1394.28	1377.44
LiF	910.60		891.62	892.88	892.81	893.16	905.41	906.59	904.77	894.04
LiCl	643.00		638.40	639.70	639.72	640.09	649.50	650.72	642.66	644.19
LiO	814.60		798.41	798.95	798.89	799.31	813.86	813.97	810.73	797.92
LiNa	256.80		254.54	249.03	249.29	249.62	251.78	245.86	246.33	245.92
Be ₂	267.90		339.05	323.64	323.90	327.12	351.44	352.32	327.20	347.20
BeH	2060.80		1985.19	1984.83	1984.70	1984.22	2040.86	2040.48	2009.73	1974.63
BeH ⁺	2221.70		2146.42	2149.42	2149.13	2148.95	2184.24	2186.90	2203.03	2109.72
BeF	1247.40		1203.54	1206.17	1206.13	1206.60	1211.30	1213.90	1222.55	1196.15
BeCl	846.70		812.04	812.50	812.56	813.26	821.58	821.89	816.44	809.36
BeO	1487.30		1463.82	1468.96	1468.96	1470.20	1474.61	1479.45	1508.31	1455.69
BeS	997.90		984.82	988.50	988.72	989.85	991.92	995.86	1008.22	975.83
B ₂	1051.30		997.04	1008.78	1010.37	1009.68	997.94	1013.61	1012.77	1011.60
BH	2366.90		2274.37	2272.63	2273.02	2270.34	2347.26	2346.24	2301.01	2245.71
BF	1402.10		1371.26	1374.12	1373.97	1374.96	1381.29	1383.48	1398.46	1354.73
BCl	840.30		823.70	816.39	816.44	817.46	839.54	831.12	826.55	810.20
BN	1514.60		1534.84	1524.21	1524.17	1526.35	1556.10	1544.90	1570.16	1513.24
BO	1885.70		1853.12	1854.69	1854.60	1856.23	1867.82	1868.82	1896.33	1844.00
BS	1180.20		1170.12	1169.79	1170.51	1171.89	1180.66	1180.60	1195.46	1161.49
C ₂	1854.70		1846.34	1857.32	1860.38	1862.84	1845.50	1860.90	1891.01	1857.11
CH	2858.50		2738.12	2725.24	2726.40	2727.85	2817.65	2808.77	2803.97	2730.18
CF	1308.10		1254.54	1262.65	1263.46	1264.39	1258.56	1266.11	1297.42	1257.32
CN	2068.60		2093.37	2096.91	2098.35	2100.44	2105.00	2108.94	2149.60	2093.57
CO	2169.80		2153.95	2148.45	2148.72	2150.60	2167.73	2162.13	2195.55	2129.62
CO ⁺	2214.20		2209.94	2213.52	2214.18	2216.64	2224.73	2228.25	2277.95	2202.58
CP	1239.70		1262.02	1263.72	1264.31	1265.84	1270.35	1271.96	1286.41	1250.18
CS	1285.20		1285.03	1276.40	1277.34	1278.70	1299.05	1290.86	1304.75	1263.69
N ₂	2358.60		2384.46	2373.02	2374.48	2376.59	2402.25	2391.08	2432.68	2353.66
N ₂ ⁺	2207.00		2263.01	2267.89	2270.16	2272.37	2272.60	2279.75	2326.82	2264.91
NH	3282.30		3176.96	3182.65	3183.72	3178.92	3242.12	3247.86	3246.50	3171.82
NF	1141.40		1115.21	1136.97	1137.14	1137.98	1116.17	1137.95	1171.33	1128.53
NCl	828.00		818.88	824.79	826.06	826.57	818.83	825.33	836.47	835.85
NO	1904.20		1897.34	1899.99	1901.66	1903.83	1907.39	1911.17	1959.47	1892.16
NO ⁺	2376.40		2395.45	2387.78	2389.33	2391.71	2413.75	2406.85	2456.14	2363.83
NS	1218.70		1219.33	1222.56	1223.24	1224.76	1227.34	1230.68	1254.42	1213.17
O ₂	1580.20		1554.74	1563.71	1565.23	1566.68	1557.62	1568.15	1624.96	1555.90
O ₂ ⁺	1904.80		1941.06	1947.67	1949.39	1951.56	1949.82	1958.07	2026.58	1931.33
OH	3737.80		3627.98	3622.10	3622.55	3627.47	3654.09	3648.88	3718.29	3602.70
OH ⁺	3113.40		2995.45	2992.68	2992.91	2994.57	3056.99	3054.84	3060.89	2953.36
F ₂	916.60		1002.19	1007.52	1006.95	1006.40	1003.89	1007.99	1048.14	997.29
F ₂ ⁺	1073.30		1161.56	1167.18	1166.73	1167.64	1163.55	1168.53	1219.83	1151.83
HF	4138.30		3981.32	3992.67	3992.60	3997.21	3994.59	4005.79	4093.34	3977.20
HF ⁺	3090.50		2950.45	2945.42	2945.42	2949.03	2989.28	2983.36	3024.75	2910.03
Na ₂	159.10		158.26	159.67	159.77	159.82	149.71	152.12	155.29	158.96
NaH	1172.20		1158.49	1157.40	1157.37	1157.21	1177.48	1178.33	1167.88	1143.28
NaF	535.70		532.82	533.51	533.49	533.71	533.88	534.62	539.36	519.85

(continued)

NaO	492.30	481.58	481.60	481.59	481.87	484.27	484.31	488.05	466.16
MgH	1495.20	1426.67	1427.09	1426.83	1426.73	1491.74	1491.76	1461.59	1409.21
MgH ⁺	1699.10	1681.16	1681.93	1681.72	1681.62	1711.13	1711.70	1720.88	1642.69
MgO	784.80	806.52	817.18	817.05	817.70	805.59	815.84	837.42	803.21
MgS	528.70	534.86	541.36	541.32	541.94	534.79	541.67	550.24	526.22
Al ₂	350.00	352.59	354.54	355.01	355.49	352.68	356.04	355.77	341.93
AlH	1682.60	1627.00	1629.99	1629.37	1628.22	1680.51	1682.04	1646.51	1585.99
AlF	802.30	769.35	771.59	771.53	772.08	774.87	776.90	788.75	747.66
AlCl	481.30	469.23	471.00	470.88	471.32	474.81	476.30	475.87	458.80
AlO	979.20	951.23	964.97	964.59	965.60	951.57	965.69	992.26	956.88
AlS	617.10	606.76	614.05	614.07	615.03	610.28	618.05	630.55	600.43
Si ₂	511.00	554.83	553.56	554.00	554.97	559.61	558.72	564.20	483.24
SiH	2041.80	1981.19	1984.29	1983.17	1982.79	2041.66	2044.43	2020.24	1951.71
SiH ⁺	2157.20	2093.81	2098.33	2097.95	2096.81	2140.07	2144.42	2119.64	2047.83
SiF	857.20	825.32	829.84	829.54	830.28	829.21	834.03	847.30	807.65
SiCl	535.60	520.92	523.51	523.20	523.83	524.34	527.70	529.27	515.10
SiN	1151.40	1160.86	1165.03	1164.84	1165.80	1165.76	1169.33	1179.68	1150.13
SiO	1241.50	1217.05	1219.35	1219.16	1220.13	1225.91	1227.77	1247.89	1196.56
SiS	749.60	741.01	741.35	741.53	742.31	749.07	749.27	753.33	726.42
P ₂	780.80	798.51	797.35	797.87	798.72	808.14	805.83	813.84	782.42
P ₂ ⁺	672.20	693.34	690.24	690.94	691.85	703.11	700.45	707.78	674.31
PH	2365.20	2313.07	2314.07	2314.35	2314.98	2369.48	2368.74	2367.90	2288.59
PF	846.80	822.79	828.46	827.94	828.66	826.47	831.54	846.90	811.02
PCl	551.40	540.49	542.78	542.79	543.19	542.39	544.09	550.12	538.06
PN	1337.20	1361.57	1358.90	1358.82	1360.18	1373.75	1370.51	1390.71	1339.51
PO	1233.30	1214.19	1219.85	1219.51	1220.80	1221.97	1227.10	1250.59	1200.39
S ₂	725.60	709.90	713.89	714.98	715.74	713.63	718.56	732.65	701.88
SO	1149.20	1117.20	1124.33	1124.27	1125.55	1121.43	1128.30	1159.28	1110.40
Cl ₂	559.70	538.45	541.10	541.26	541.59	539.92	542.98	555.49	540.40
Cl ₂ ⁺	645.60	626.46	629.64	630.10	630.62	627.93	631.30	649.26	624.21
HCl	2990.90	2937.18	2920.15	2922.44	2926.04	2969.46	2953.90	2981.86	2901.24
HCl ⁺	2673.70	2600.18	2581.42	2583.30	2586.47	2653.39	2635.95	2643.90	2571.65
ClF	786.10	760.96	764.79	764.77	765.21	762.46	766.26	791.24	760.84
ClO	853.80	841.27	849.04	849.56	850.23	836.64	844.85	872.53	862.59
ME		-21.00	-19.62	-19.28	-18.35	-4.67	-2.87	11.34	-33.78
MAD		35.57	34.28	34.27	34.12	25.18	24.14	30.90	43.61

VI. SOLID TEST SET RESULTS

Here we provide the solid-by-solid tabulation for the band gaps (21 systems), equilibrium lattice constants and cohesive energies (55 solids), and bulk moduli (44 cubic solids) for *rreg*TM compared with *sreg*TM, r^2 SCAN, and PBE.

TABLE XII. Band Gap (eV) of 21 insulators and semiconductors. Experimental lattice parameters were used with all functionals. Experimental values Exptl. and lattice constants are from Ref. [12].

Solid	Exp.	X C	<i>rreg</i> TM <i>rreg</i> TM	<i>v1-sreg</i> TM <i>rreg</i> TM	<i>v2-sreg</i> TM <i>rreg</i> TM	<i>v3-sreg</i> TM <i>rreg</i> TM	<i>rreg</i> TM <i>PBE</i>	<i>v2-sreg</i> TM <i>PBE</i>	r^2 SCAN r^2 SCAN	<i>PBE</i> <i>PBE</i>
C	5.50		4.48	4.13	4.14	4.16	4.76	4.42	4.32	4.13
Si	1.17		0.82	0.63	0.63	0.65	1.06	0.88	0.78	0.58
Ge	0.74		0.33	0.38	0.39	0.41	0.37	0.44	0.50	0.09
SiC	2.42		1.52	1.44	1.46	1.48	1.61	1.54	1.75	1.38
BN	6.36		4.72	4.57	4.59	4.62	4.87	4.73	5.00	4.48
BP	2.10		1.50	1.28	1.28	1.30	1.71	1.50	1.45	1.24
AlN	4.90		3.46	3.48	3.50	3.52	3.49	3.52	3.97	3.35
AlP	2.50		1.79	1.70	1.71	1.73	1.93	1.84	1.92	1.59
AlAs	2.23		1.63	1.59	1.60	1.61	1.70	1.66	1.80	1.44
GaN	3.28		1.69	1.76	1.77	1.79	1.68	1.76	2.19	1.74
GaP	2.35		1.68	1.66	1.67	1.69	1.76	1.73	1.89	1.62
GaAs	1.52		0.93	1.00	1.00	1.02	0.99	1.07	1.09	0.61
InP	1.42		0.96	0.99	0.99	1.01	1.04	1.07	1.18	0.75
InAs	0.42		0.11	0.17	0.18	0.19	0.15	0.22	0.25	0.00
InSb	0.24		0.18	0.23	0.24	0.25	0.22	0.28	0.27	0.00
LiH	4.94		3.34	3.33	3.34	3.34	3.69	3.69	3.72	3.07
LiF	14.20		9.24	9.30	9.31	9.32	9.21	9.28	10.19	9.20
LiCl	9.40		6.77	6.78	6.79	6.80	6.85	6.87	7.31	6.41
NaF	11.50		6.25	6.32	6.32	6.33	6.25	6.33	7.21	6.37
NaCl	8.50		5.36	5.39	5.39	5.41	5.44	5.48	5.98	5.19
MgO	7.83		4.98	5.04	5.05	5.07	4.98	5.05	5.65	4.77
ME			-1.52	-1.54	-1.53	-1.52	-1.42	-1.44	-1.20	-1.69
MAD			1.52	1.54	1.53	1.52	1.42	1.44	1.20	1.69

TABLE XIII. Equilibrium lattice constants, $a_0(\text{\AA})$ comparisons of 55 solids. Experimental values Exptl. are from Ref. [10], include zero-point effects.

Solid	Exp.	X C	<i>rreg</i> TM <i>rreg</i> TM	<i>v1-sreg</i> TM <i>rreg</i> TM	<i>v2-sreg</i> TM <i>rreg</i> TM	<i>v3-sreg</i> TM <i>rreg</i> TM	<i>rreg</i> TM <i>PBE</i>	<i>v2-sreg</i> TM <i>PBE</i>	r ² SCAN r ² SCAN	<i>PBE</i> <i>PBE</i>
C	3.553		3.560	3.565	3.565	3.565	3.555	3.560	3.562	3.572
Si	5.421		5.425	5.430	5.430	5.430	5.417	5.420	5.440	5.469
Ge	5.644		5.693	5.698	5.700	5.701	5.680	5.687	5.681	5.782
Sn	6.477		6.545	6.555	6.559	6.561	6.528	6.542	6.565	6.653
SiC	4.346		4.356	4.352	4.353	4.352	4.352	4.347	4.355	4.379
BN	3.592		3.615	3.617	3.618	3.617	3.609	3.612	3.614	3.626
BP	4.525		4.526	4.530	4.530	4.530	4.521	4.523	4.533	4.547
AlN	4.368		4.374	4.372	4.372	4.371	4.367	4.366	4.368	4.402
AlP	5.451		5.463	5.465	5.466	5.465	5.454	5.456	5.472	5.506
AlAs	5.649		5.674	5.672	5.674	5.674	5.667	5.665	5.677	5.733
GaN	4.520		4.560	4.557	4.557	4.557	4.555	4.552	4.527	4.589
GaP	5.439		5.487	5.487	5.487	5.487	5.480	5.479	5.467	5.534
GaAs	5.640		5.684	5.688	5.690	5.691	5.673	5.679	5.672	5.762
InP	5.858		5.951	5.949	5.950	5.950	5.943	5.941	5.947	6.001
InAs	6.047		6.128	6.132	6.135	6.136	6.115	6.122	6.131	6.211
InSb	6.468		6.548	6.558	6.561	6.563	6.533	6.546	6.569	6.647
LiH	3.979		3.983	3.984	3.986	3.986	3.940	3.944	4.012	4.016
LiF	3.972		3.990	3.992	3.993	3.992	3.969	3.971	3.994	4.071
LiCl	5.070		5.078	5.080	5.082	5.081	5.044	5.047	5.111	5.153
NaF	4.582		4.497	4.501	4.501	4.501	4.491	4.495	4.503	4.632
NaCl	5.569		5.504	5.508	5.509	5.509	5.489	5.495	5.545	5.655
MgO	4.189		4.209	4.206	4.207	4.205	4.202	4.200	4.196	4.248
Li	3.443		3.432	3.433	3.434	3.434	3.401	3.403	3.479	3.439
Na	4.214		4.113	4.117	4.119	4.120	4.111	4.118	4.200	4.193
K	5.212		5.199	5.203	5.207	5.209	5.202	5.213	5.350	5.286
Rb	5.577		5.546	5.554	5.559	5.560	5.572	5.585	5.751	5.669
Cs	6.039		6.044	6.057	6.063	6.067	6.063	6.083	6.271	6.161
Ca	5.556		5.484	5.473	5.475	5.474	5.484	5.482	5.574	5.525
Ba	5.002		4.988	4.994	4.999	5.002	4.990	5.002	5.077	5.030
Sr	6.040		5.987	5.984	5.988	5.988	5.989	5.994	6.101	6.027
Al	4.018		3.986	3.983	3.982	3.981	3.980	3.975	3.988	4.040
Fe	2.853		2.799	2.805	2.807	2.808	2.798	2.806	2.865	2.829
Co	3.524		3.501	3.494	3.497	3.498	3.502	3.500	3.514	3.517
Ni	3.508		3.476	3.477	3.479	3.480	3.470	3.473	3.479	3.518
Sc	3.270		3.235	3.233	3.235	3.235	3.234	3.234	3.260	3.249
Y	3.594		3.588	3.591	3.593	3.594	3.587	3.595	3.623	3.592
Ti	2.915		2.878	2.877	2.879	2.879	2.876	2.877	2.887	2.893
Zr	3.198		3.205	3.203	3.206	3.206	3.204	3.205	3.215	3.210
Hf	3.151		3.129	3.132	3.134	3.135	3.127	3.132	3.141	3.158
V	3.021		2.964	2.964	2.966	2.966	2.962	2.963	2.965	2.978
Nb	3.294		3.319	3.316	3.318	3.319	3.317	3.317	3.320	3.322
Ta	3.299		3.283	3.285	3.287	3.288	3.280	3.284	3.290	3.309
Mo	3.141		3.139	3.139	3.141	3.141	3.136	3.138	3.144	3.151
W	3.160		3.151	3.152	3.154	3.154	3.148	3.151	3.156	3.172
Tc	2.716		2.711	2.711	2.713	2.713	2.708	2.710	2.716	2.725
Re	2.744		2.746	2.747	2.749	2.749	2.744	2.747	2.750	2.764
Ru	2.669		2.669	2.671	2.673	2.673	2.667	2.670	2.672	2.683
Os	2.699		2.703	2.705	2.706	2.707	2.701	2.704	2.708	2.723
Rh	3.794		3.800	3.803	3.806	3.806	3.796	3.801	3.804	3.824
Ir	3.831		3.844	3.848	3.850	3.851	3.840	3.846	3.852	3.872
Pd	3.876		3.902	3.905	3.909	3.910	3.896	3.903	3.912	3.939
Pt	3.913		3.933	3.937	3.941	3.942	3.927	3.935	3.943	3.968
Cu	3.595		3.569	3.574	3.577	3.578	3.563	3.570	3.582	3.634
Ag	4.062		4.082	4.087	4.092	4.094	4.075	4.084	4.108	4.147
Au	4.062		4.108	4.114	4.118	4.120	4.100	4.111	4.128	4.156
ME			0.000	0.002	0.004	0.005	-0.006	-0.002	0.026	0.046
MAD			0.029	0.030	0.031	0.031	0.028	0.029	0.037	0.053

TABLE XIV. Cohesive energies, $E_{coh}(eV/atom)$ of 55 solids. Experimental values Exptl. are from Ref. [10], include zero-point effects.

Solid	Exp.	X C	<i>rreg</i> TM <i>rreg</i> TM	<i>v1-sreg</i> TM <i>rreg</i> TM	<i>v2-sreg</i> TM <i>rreg</i> TM	<i>v3-sreg</i> TM <i>rreg</i> TM	<i>rreg</i> TM <i>PBE</i>	<i>v2-sreg</i> TM <i>PBE</i>	r ² SCAN r ² SCAN	<i>PBE</i> <i>PBE</i>
C	7.550		7.552	7.524	7.528	7.539	7.567	7.527	7.467	7.714
Si	4.680		4.671	4.640	4.642	4.650	4.746	4.696	4.675	4.561
Ge	3.890		3.923	3.892	3.889	3.883	3.946	3.912	3.891	3.725
Sn	3.160		3.409	3.360	3.356	3.348	3.431	3.377	3.295	3.200
SiC	6.480		6.423	6.468	6.462	6.470	6.434	6.465	6.432	6.408
BN	6.760		6.919	6.912	6.906	6.914	6.916	6.894	6.788	6.936
BP	5.140		5.339	5.275	5.277	5.287	5.383	5.302	5.281	5.293
AlN	5.850		5.891	5.930	5.922	5.926	5.890	5.920	5.809	5.711
AlP	4.320		4.287	4.275	4.270	4.276	4.337	4.308	4.249	4.093
AlAs	3.820		3.886	3.879	3.874	3.875	3.903	3.886	3.867	3.693
GaN	4.550		4.615	4.628	4.620	4.619	4.633	4.637	4.465	4.385
GaP	3.610		3.698	3.690	3.685	3.685	3.732	3.715	3.642	3.478
GaAs	3.340		3.406	3.376	3.372	3.366	3.429	3.395	3.328	3.161
InP	3.470		3.358	3.354	3.348	3.346	3.382	3.371	3.245	3.123
InAs	3.080		3.154	3.119	3.114	3.108	3.175	3.135	3.026	2.893
InSb	2.810		2.881	2.835	2.830	2.823	2.901	2.850	2.729	2.650
LiH	2.490		2.450	2.449	2.447	2.446	2.474	2.470	2.420	2.348
LiF	4.460		4.339	4.391	4.388	4.390	4.290	4.339	4.395	4.321
LiCl	3.590		3.482	3.518	3.516	3.517	3.435	3.469	3.467	3.352
NaF	3.970		3.923	3.973	3.970	3.971	3.880	3.927	3.992	3.860
NaCl	3.340		3.258	3.293	3.291	3.292	3.228	3.260	3.260	3.088
MgO	5.200		5.139	5.183	5.177	5.182	5.155	5.192	5.249	4.998
Li	1.670		1.638	1.641	1.641	1.641	1.591	1.595	1.569	1.600
Na	1.120		1.140	1.137	1.134	1.133	1.124	1.118	1.041	1.038
K	0.940		0.959	0.957	0.954	0.953	0.958	0.952	0.841	0.857
Rb	0.860		0.858	0.853	0.849	0.847	0.877	0.868	0.734	0.760
Cs	0.810		0.813	0.807	0.803	0.801	0.821	0.812	0.661	0.693
Ca	1.870		1.898	1.930	1.923	1.920	2.025	2.052	1.917	1.846
Ba	1.910		1.968	1.958	1.947	1.944	2.099	2.078	1.817	1.703
Sr	1.730		1.855	1.843	1.835	1.833	2.000	1.981	1.786	1.604
Al	3.430		3.653	3.662	3.656	3.652	3.662	3.667	3.609	3.442
Fe	4.300		5.064	5.153	5.121	5.100	5.102	5.166	4.635	4.871
Co	4.420		5.023	5.269	5.242	5.214	5.055	5.360	4.871	5.165
Ni	4.480		5.127	5.392	5.132	5.105	5.438	5.193	4.645	4.732
Sc	3.930		4.277	4.349	4.337	4.329	4.635	4.410	4.017	4.142
Y	4.390		4.405	4.402	4.386	4.377	4.523	4.505	4.014	4.100
Ti	4.880		5.158	5.381	5.364	5.353	5.166	5.372	5.060	5.442
Zr	6.270		6.322	6.007	6.001	5.940	6.376	6.049	4.378	5.344
Hf	6.460		6.948	6.880	6.854	6.838	7.087	6.994	6.434	6.455
V	5.350		6.331	5.637	5.606	5.583	5.654	5.623	4.660	6.520
Nb	7.600		7.269	7.205	7.168	7.146	7.299	7.204	6.361	6.919
Ta	8.130		8.952	8.845	8.807	8.783	9.087	8.942	8.251	8.314
Mo	6.860		6.830	6.721	6.676	6.648	6.875	6.720	5.690	6.351
W	8.940		9.205	9.052	9.017	8.985	9.279	9.090	8.067	8.485
Tc	6.880		7.479	7.353	7.299	7.323	7.502	7.360	6.337	6.971
Re	8.050		8.654	8.490	8.440	8.403	8.783	8.568	7.514	7.815
Ru	6.770		6.997	7.005	6.954	6.922	7.028	6.987	6.338	6.773
Os	8.200		9.099	8.926	8.872	8.834	9.223	8.994	8.072	8.356
Rh	5.780		6.116	6.096	6.050	6.018	6.150	6.085	5.826	5.748
Ir	6.990		8.002	7.832	7.783	7.745	8.050	7.851	7.142	7.494
Pd	3.930		4.596	4.503	4.470	4.444	4.652	4.524	4.163	3.738
Pt	5.870		6.303	6.172	6.130	6.101	6.386	6.212	5.716	5.531
Cu	3.510		4.306	4.227	4.198	4.181	4.370	4.262	3.855	3.479
Ag	2.960		3.261	3.193	3.168	3.152	3.310	3.217	2.873	2.511
Au	3.830		3.820	3.730	3.702	3.681	3.883	3.764	3.417	3.033
ME			0.212	0.18	0.159	0.148	0.248	0.199	-0.134	-0.070
MAD			0.251	0.232	0.216	0.210	0.288	0.251	0.238	0.252

TABLE XV. Bulk modulus, $B_0(GPa)$ of 44 cubic solids. Experimental values Exptl. are from Ref. [11], these values were obtained by subtracting the zero-point phonon effect from the experimental zero-temperature values.

Solid	Exp.	X C	$rregTM$ $rregTM$	$v1-sregTM$ $rregTM$	$v2-sregTM$ $rregTM$	$v3-sregTM$ $rregTM$	$rregTM$ PBE	$v2-sregTM$ PBE	r^2SCAN r^2SCAN	PBE PBE
C	454.700		449.794	439.039	439.805	440.565	459.618	449.531	449.244	433.061
Si	101.300		98.870	95.336	95.389	95.712	103.557	100.463	97.272	88.798
Ge	79.400		70.014	69.709	69.611	69.619	72.193	71.784	73.095	59.083
Sn	42.800		43.263	42.780	42.673	42.617	44.541	43.929	43.526	36.172
SiC	229.100		225.815	224.577	224.220	224.633	229.867	228.388	226.927	212.361
BN	410.200		387.268	383.780	383.296	384.044	394.985	390.743	389.901	372.412
BP	168.000		171.053	166.161	166.357	166.754	176.357	171.680	170.052	161.851
AlN	206.000		206.487	207.245	207.037	207.479	209.598	210.108	212.380	193.670
AlP	87.400		91.687	90.035	89.942	90.195	94.901	93.076	91.095	82.678
AlAs	75.000		76.308	75.745	75.557	75.670	78.332	77.566	76.387	67.252
GaN	213.000		197.682	198.807	198.521	198.914	200.704	201.514	211.373	182.466
GaP	89.600		88.431	87.500	87.463	87.681	90.981	89.785	90.466	78.712
GaAs	76.700		72.167	72.145	71.990	72.014	73.933	73.764	73.804	62.019
InP	72.000		68.468	68.339	68.256	68.405	70.189	69.924	70.286	60.930
InAs	58.600		57.540	57.441	57.289	57.287	58.911	58.661	58.929	49.497
InSb	46.100		43.781	43.336	43.221	43.163	44.879	44.299	44.017	37.074
LiH	40.100		37.627	37.545	37.481	37.453	40.922	40.751	36.485	36.343
LiF	76.300		79.448	79.030	78.996	79.054	83.075	82.599	78.714	67.393
LiCl	38.700		36.962	36.730	36.717	36.723	39.062	38.798	35.064	31.863
NaF	53.100		60.176	59.892	59.873	59.867	60.462	60.157	59.895	47.016
NaCl	27.600		31.288	31.073	31.063	31.069	31.723	31.492	29.140	24.264
MgO	169.800		165.096	166.031	165.926	166.215	167.093	167.910	170.619	151.596
Li	13.100		13.886	13.891	13.909	13.913	14.638	14.662	13.116	13.957
Na	7.900		8.845	8.817	8.807	8.803	8.869	8.830	7.930	7.877
K	3.800		4.144	4.203	4.196	4.197	3.956	3.943	3.422	3.593
Rb	3.600		3.334	3.441	3.421	3.312	3.286	3.256	2.631	2.798
Cs	2.300		2.216	2.197	2.187	2.181	2.180	2.151	1.867	1.960
Ca	15.900		20.198	20.729	20.742	20.696	19.295	18.814	17.595	17.458
Ba	10.600		9.514	9.455	9.440	9.428	9.444	9.369	8.569	8.844
Sr	12.000		13.031	12.108	12.093	12.049	13.079	13.053	11.832	11.598
Al	77.100		90.078	90.654	91.171	91.419	91.954	93.144	92.466	77.996
Ni	192.500		226.477	224.995	223.348	222.570	229.632	226.624	216.804	196.258
V	165.800		201.548	201.041	200.526	200.408	203.115	202.116	201.269	187.488
Nb	173.200		179.288	179.820	179.238	179.021	180.805	180.736	178.078	173.453
Ta	202.700		212.406	211.688	211.184	210.991	213.916	212.688	210.119	201.055
Mo	276.200		279.506	278.740	277.833	277.557	281.413	279.859	281.995	268.535
W	327.500		330.954	329.539	328.582	328.161	333.793	331.325	329.918	312.477
Rh	277.100		281.216	279.700	277.697	276.995	285.003	281.255	281.584	258.115
Ir	362.200		379.383	375.731	373.459	372.497	384.277	378.318	378.905	349.818
Pd	187.200		194.535	191.866	189.957	189.064	198.086	193.530	186.087	169.148
Pt	285.500		276.884	273.327	271.130	270.208	281.598	275.772	273.641	249.877
Cu	144.300		160.228	155.717	153.997	152.980	166.597	160.709	159.969	138.007
Ag	105.700		114.086	111.841	110.540	109.815	116.811	113.239	103.562	90.318
Au	182.000		164.391	161.247	159.380	158.459	168.574	163.612	153.795	139.596
ME			1.856	0.666	0.223	0.140	4.375	2.732	1.367	-9.704
MAD			6.740	6.811	6.602	6.450	7.200	6.542	5.963	11.022

Acknowledgement: Work supported by U.S. National Science Foundation grant DMR-1912618.

- [1] J.P. Perdew and Y. Wang, Phys. Rev. B **45**, 13244 (1992).
- [2] J. Sun, A. Ruzsinszky, and J.P. Perdew, Phys. Rev. Lett. **115**, 036402 (2015).
- [3] J. Sun, R.C. Remsing, Y. Zhang, Z. Sun, A. Ruzsinszky, H. Peng, Z. Yang, A. Paul, U. Waghmare, X. Wu, M.L. Klein, and J.P. Perdew, Nature Chemistry **8**, 831 (2016).
- [4] J. P. Perdew, K. Burke, and M. Ernzerhof Phys. Rev. Lett **77**, 3865 (1996); erratum *ibid.* **78**, 1396 (1997).
- [5] J. Tao and Yuxiang Mo, Phys. Rev. Letter. **117**, 073001 (2016).
- [6] L.A. Curtiss, K. Raghavachari, P.C. Redfern, and J.A. Pople, J. Chem. Phys. **106**, 1063 (1997).
- [7] L.A. Curtiss, P.C. Redfern, K. Raghavachari, and J.A. Pople, J. Chem. Phys. **114**, 108 (2001).
- [8] V.N. Staroverov, G.E. Scuseria, J. Tao, and J.P Perdew, J. Chem. Phys. **119** 12129 (2003).
- [9] V.N. Staroverov, G.E. Scuseria, J. Tao, and J.P Perdew, J. Chem. Phys. **121** 11507 (2004).
- [10] H. Peng, Z.-H. Yang, J.P. Perdew, and J. Sun, Phys. Rev. X **6**, 041005 (2016).
- [11] F. Tran, J. Stelzl, and P. Blaha, J. Chem. Phys. **144**, 204120 (2016).
- [12] F. Tran and P. Blaha, J. Phys. Chem. A **121**, 3318 (2017).